

Oscillations of Polymer Electrolyte Fuel Cells and the associated Water Management

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Knowledge for Tomorrow

Outline

- Discussion of segmented cell designs for current density measurements
- Investigation of oscillatory fluctuations in PEFC
- Investigation of the triggering of the “ignition” transition
- Earlier Work on Oscillations in PEFC:

J. R. Atkins, S. C. Savett, and S. E. Creager, Journal of Power Sources 128:201 (2004)

J. F. Moxley, S. Tulyani, and J. B. Benziger, Chemical Engineering Science 58:4705 (2003).

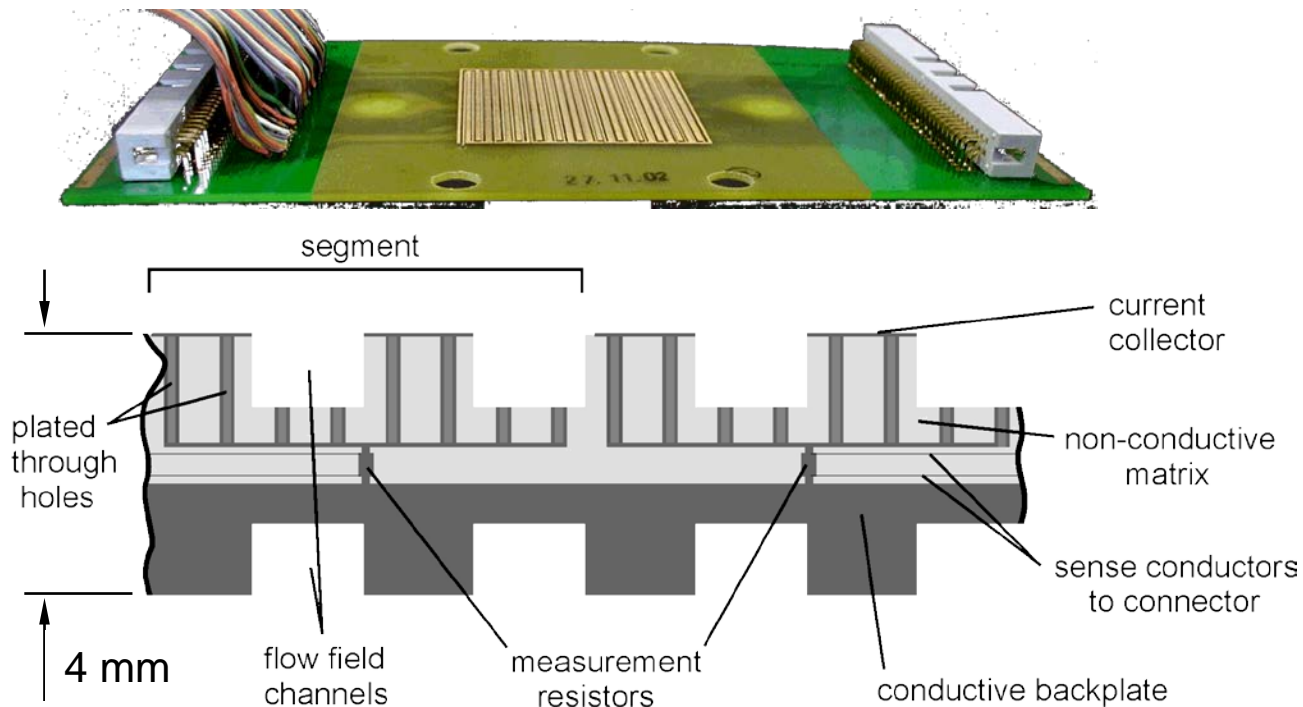
J. Benziger, E. Chia, J. F. Moxley, and I. G. Kevrekidis, Chemical Engineering Science 60:1743 (2005).

D. Garcia Sanchez, D. Diaz, R. Hiesgen, I. Wehl, and K.A. Friedrich, J. Electroanal. Chem. 649 (2010) 219-231.

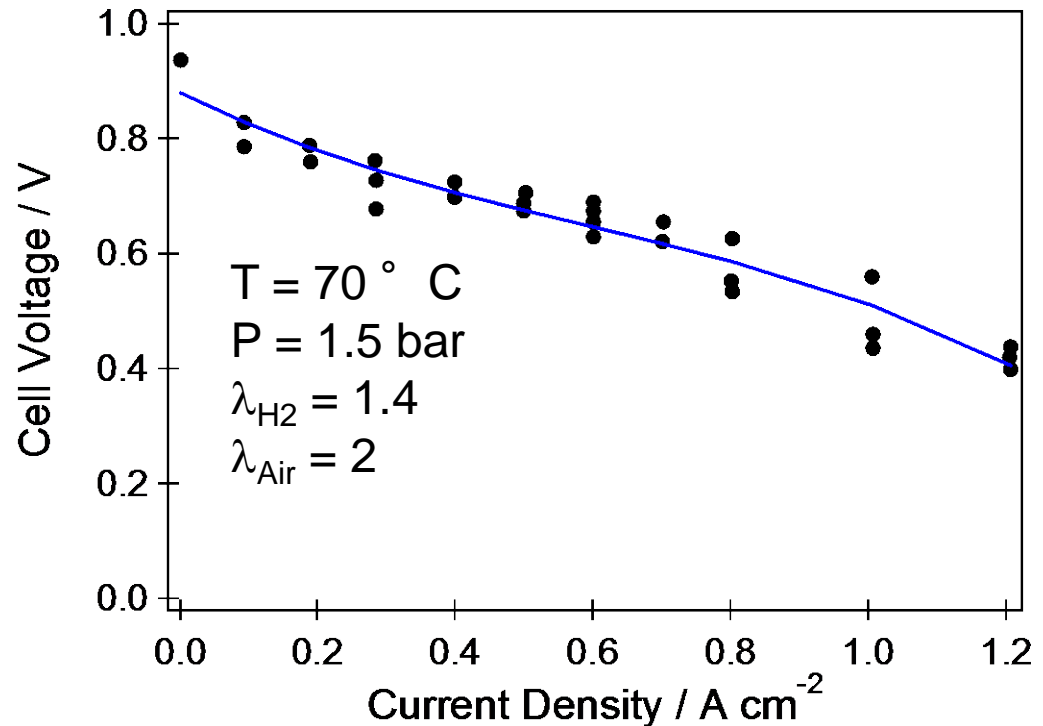
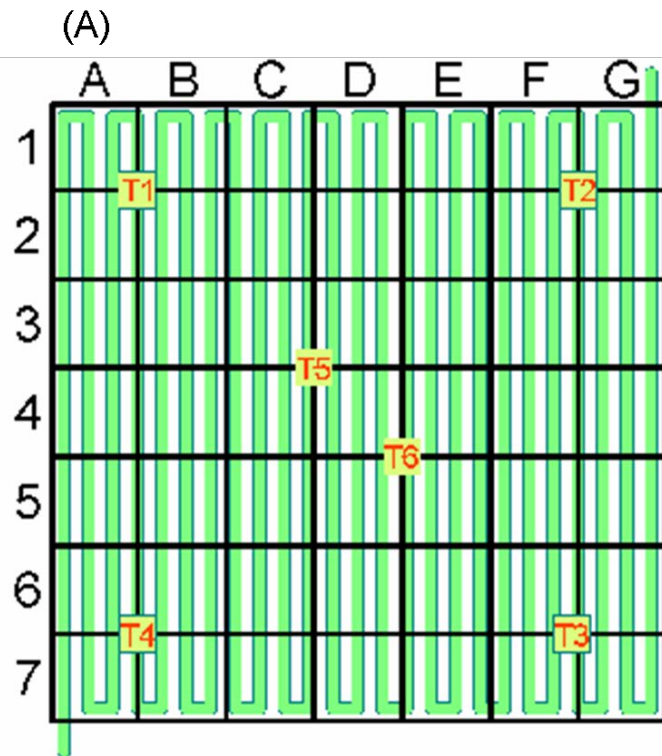


Segmented Bipolar Plate for Current Density Distribution Measurements

Current Density, Temperature, Impedance



Flow Field and Ion Power MEA Performance



Configuration **A** (co-flow):

Anode: input G1 output A7

Cathode: input G1 output A7

Configuration **B** (counter flow):

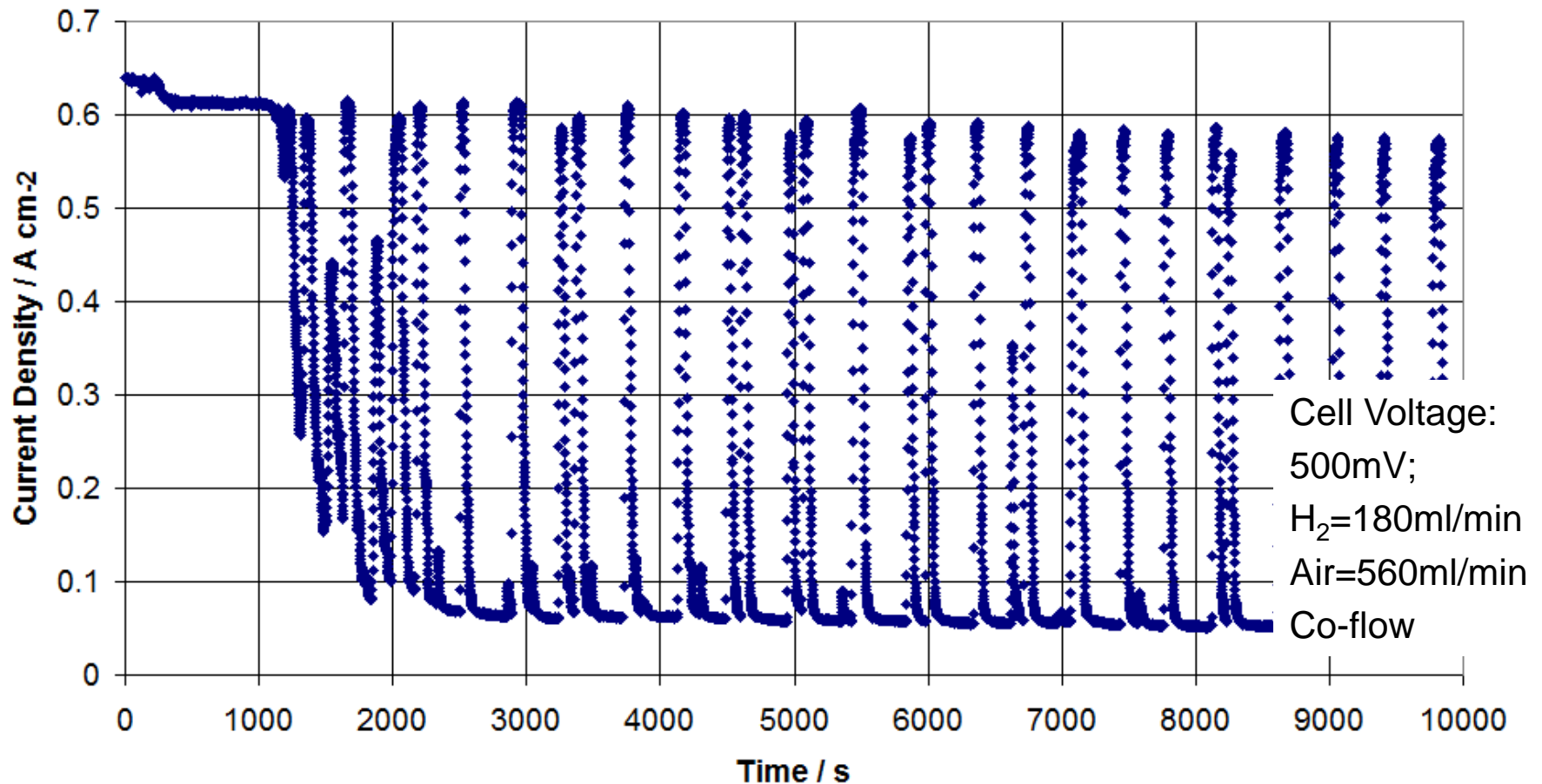
Anode: input A1 output G7

Cathode: input G1 output A7



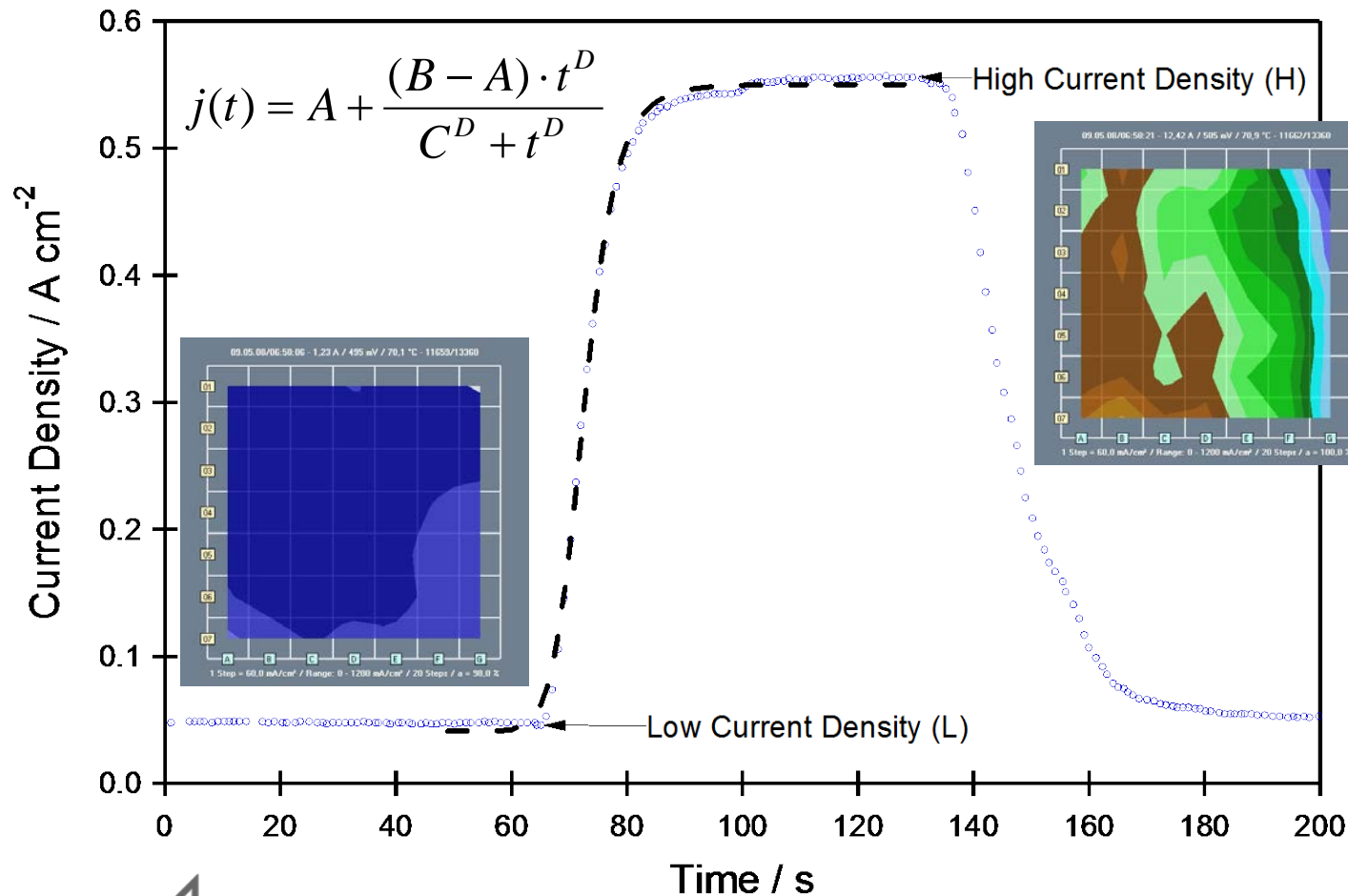
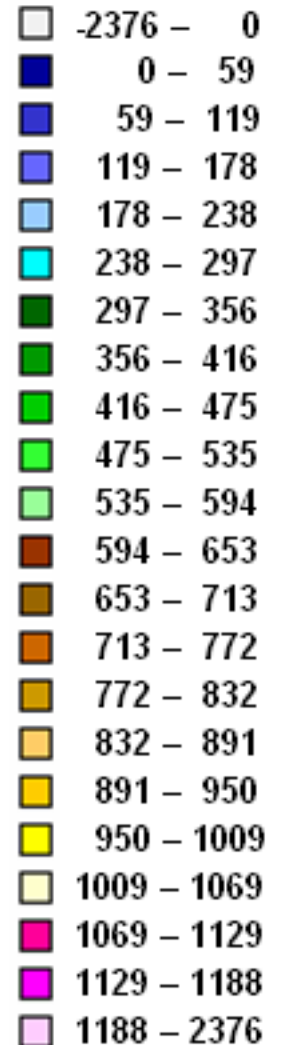
Oscillations of PEM fuel cells at low cathode humidification

Cathode with r.h.=0%, Anode with nominal r.h.=152% (condensing conditions)



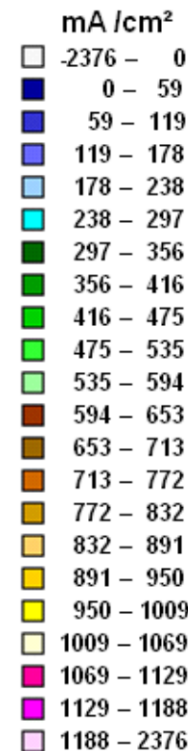
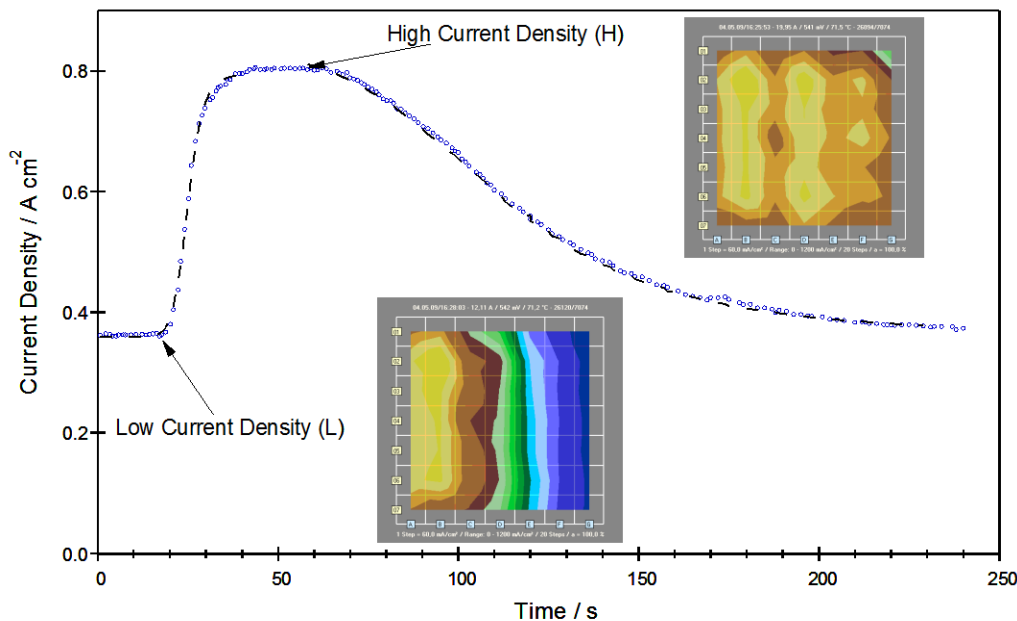
Current Density Distribution of Two Operation Points

Co-Flow

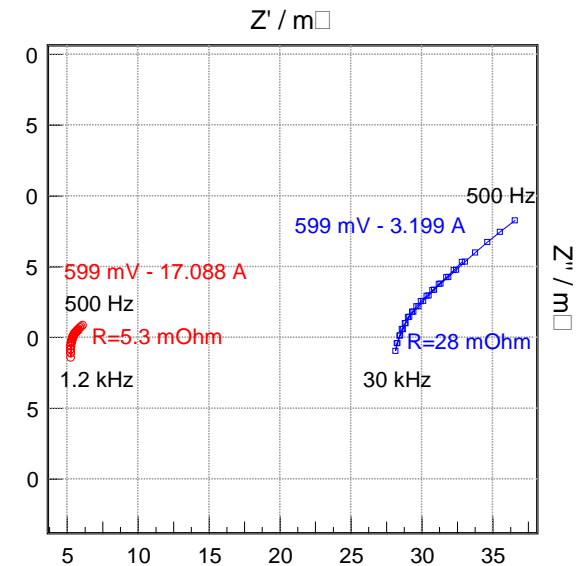
mA /cm²

Current Density Distribution of Two Operation Points

Counter Flow



Impedance spectra of two operation points

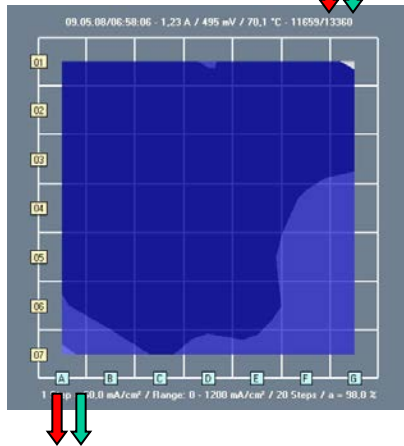


- U_{cell} : 500mV, Cathode 660ml min⁻¹, Anode 200 ml min⁻¹; T_{cell} : 70° C; counter flow

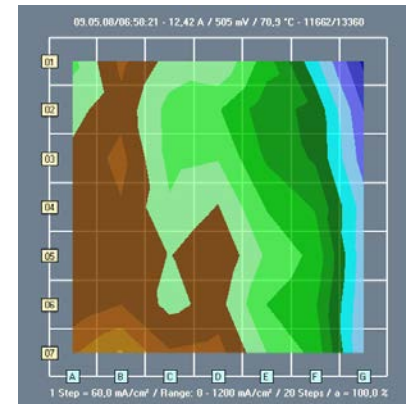
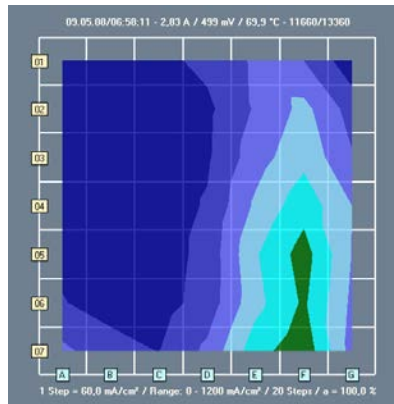


Transition from Low to High Conductivity

Anode
Cathode



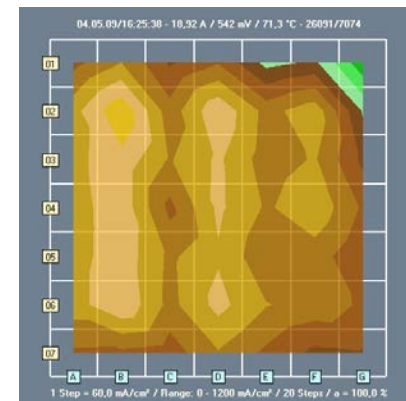
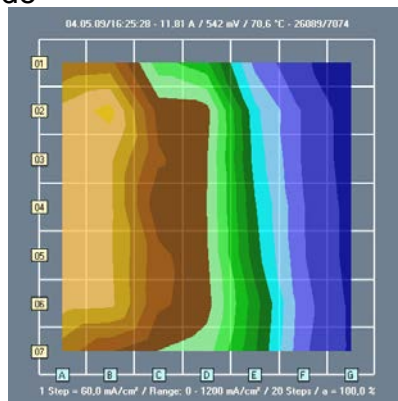
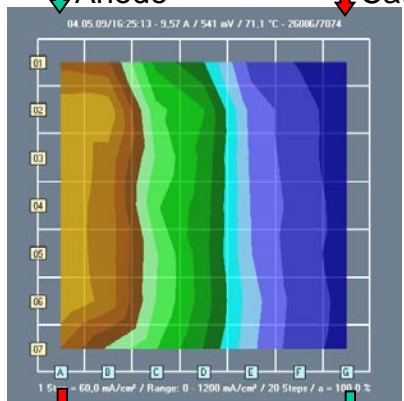
Co-Flow



Anode

Cathode

Counter Flow

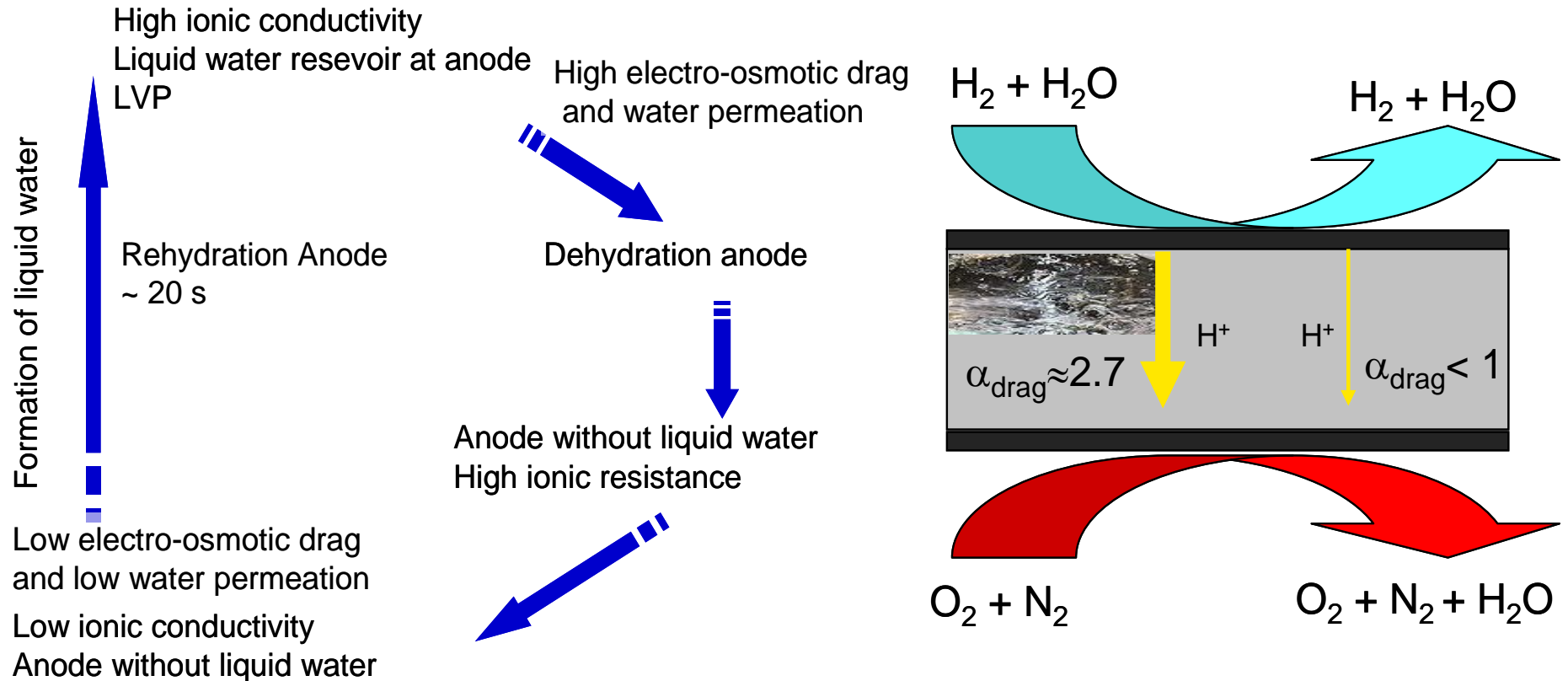


Cathode

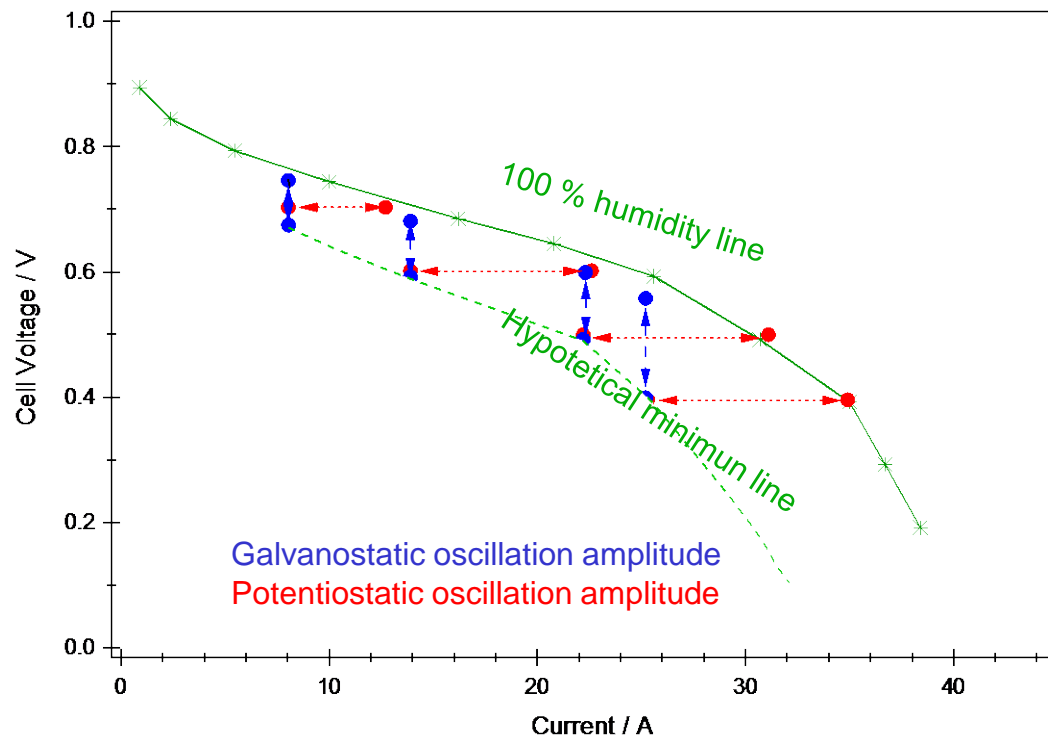
Anode



Interpretation of the Oscillation Cycle



Oscillations are present under Potentiostatic and Galvanostatic Operation Conditions



$RH_{\text{Cathode}} = 0\%$

Stoichiometric ratio: 1.5
(H_2) and 2 (Air).

Reference (25 A)

$F_{H_2} = 261.5 \text{ cm}^3/\text{min}$

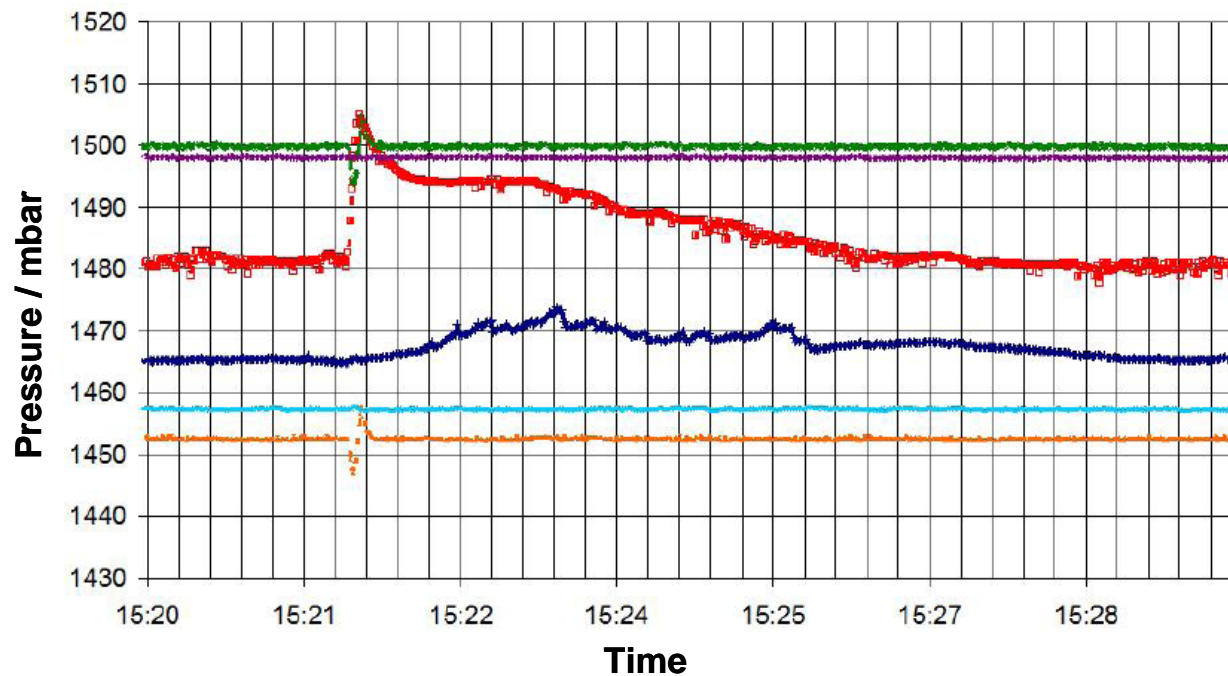
$F_{\text{Air}} = 830.9 \text{ cm}^3/\text{min}$

Galvanostatic oscillation amplitude

Potentiostatic oscillation amplitude



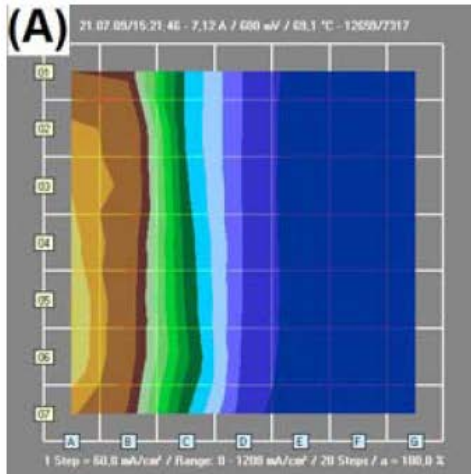
Investigation of the „Ignition“ Transition



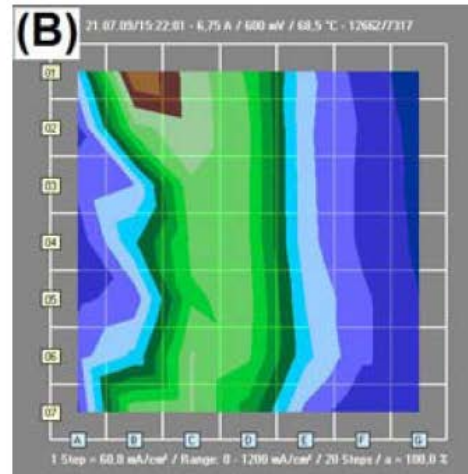
Observation of a pressure surge in the anode followed by a time delayed gradual pressure increase in the cathode compartment



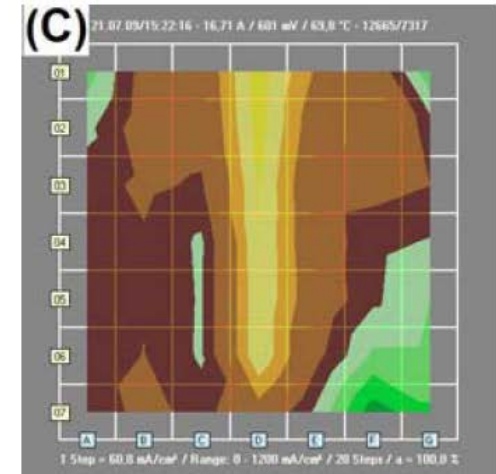
Investigation of the „Ignition“ Transition / Current Density Distributions



Low current level (L)
Current: 7.12 A



Before L to H transition
Current: 6.75 A



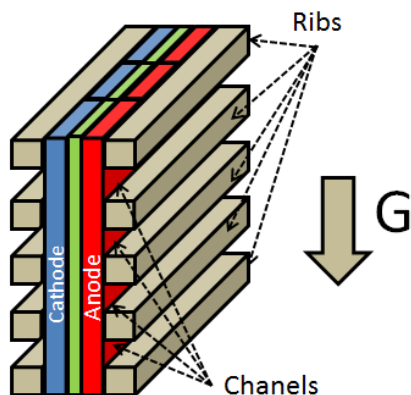
High current level (H)
Current: 16.71 A

Blockage of chanel with liquid water?

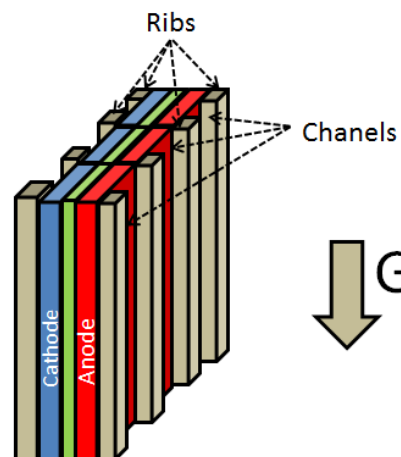


Influence of the Orientation of the Cell

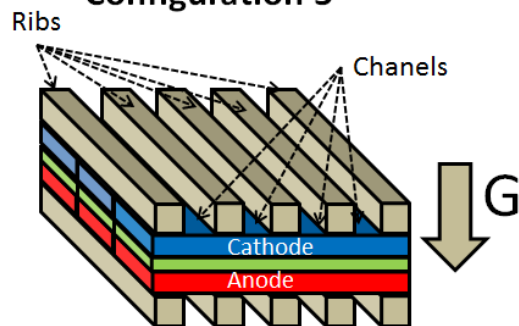
Configuration 1



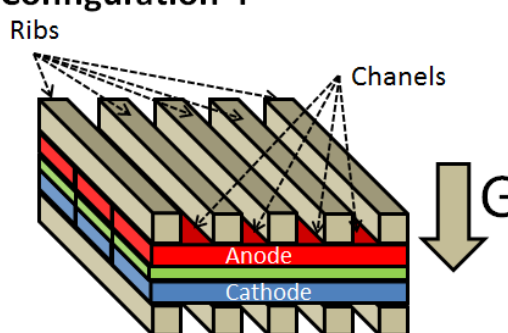
Configuration 2



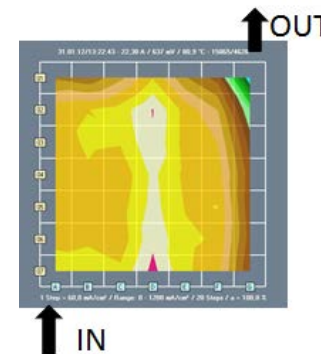
Configuration 3



Configuration 4



Co-flow



Experimental conditions:

Potentiostatic mode: 600mV

Flows:

Anode: 209ml/min

Cathode: 664ml/min

Bubbler Anode temperature: 80° C

Cell temperatures : 70° C, 80° C, 85° C

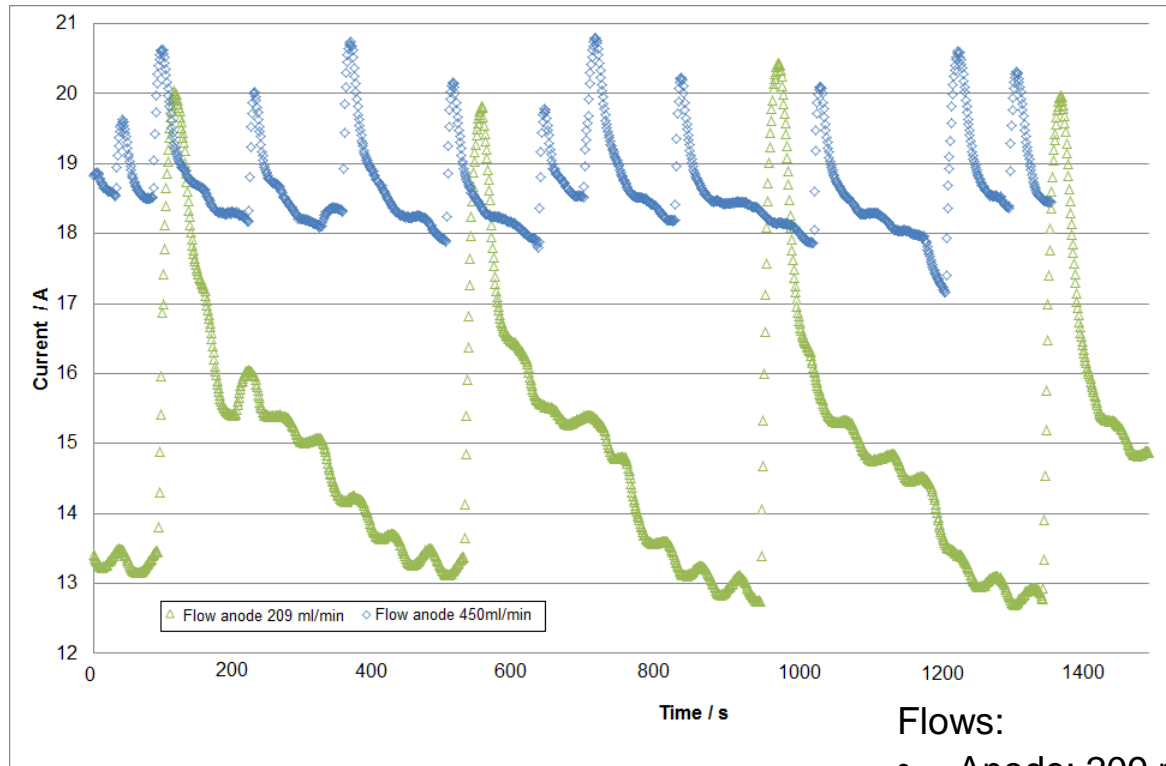
Pressure: 1.5 bar

Relativity humidity cathode: 0%

Relativity humidity anode: variable



Oscillation Frequency dependent on the Anode Flow

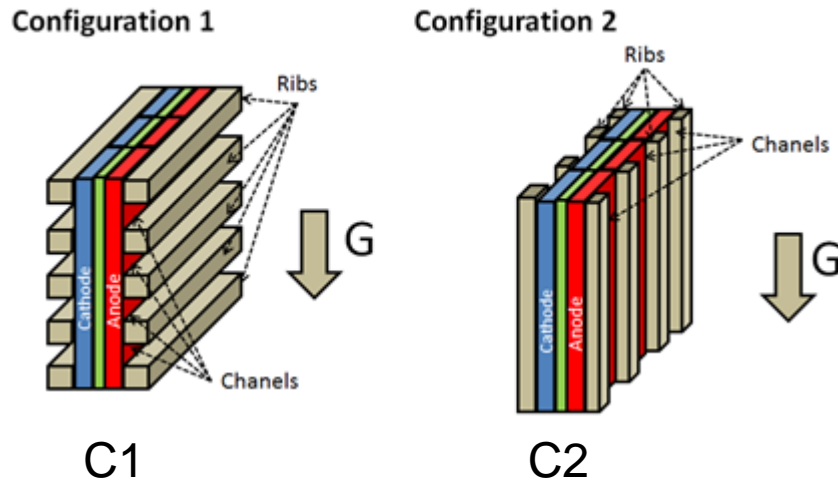


Flows:

- Anode: 209 ml/min, 450 ml/min
- Cathode: 664 ml/min
- Relativity humidity cathode: 0%
- Relativity humidity anode: 100%



Comparison of Configuration 1 and 2

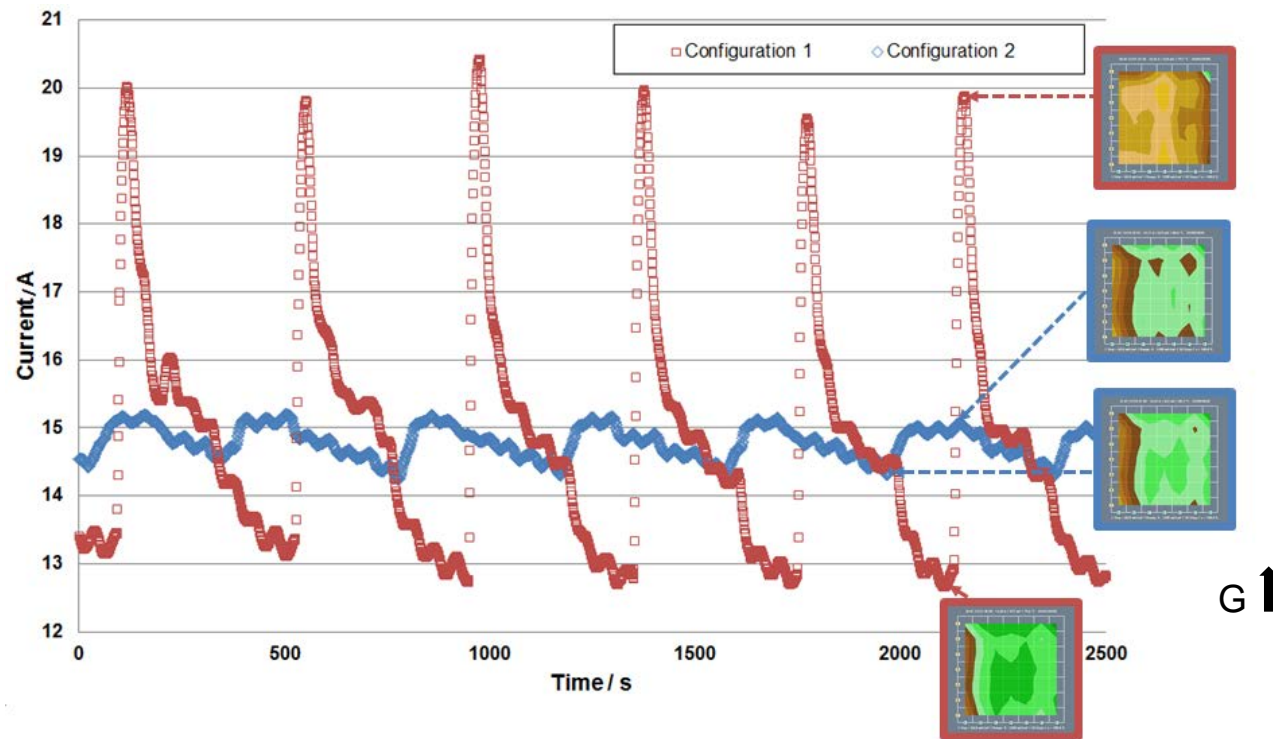


- Preferential formation of water drops observed in configuration 1 **(1)**
- Differences in the oscillatory behavior expected

(1) Zijie Lu, Cody Rath, Guangsheng Zhang, Satish G. Kandlikar. "Water management studies in PEM fuelcells, part IV: Effects of channel surface wettability, geometry and orientation on the two phase flow in parallel gas channels." *Journal of Hydrogen Energy*, 2011: 9864-9875.



Comparison of Configuration 1 and 2

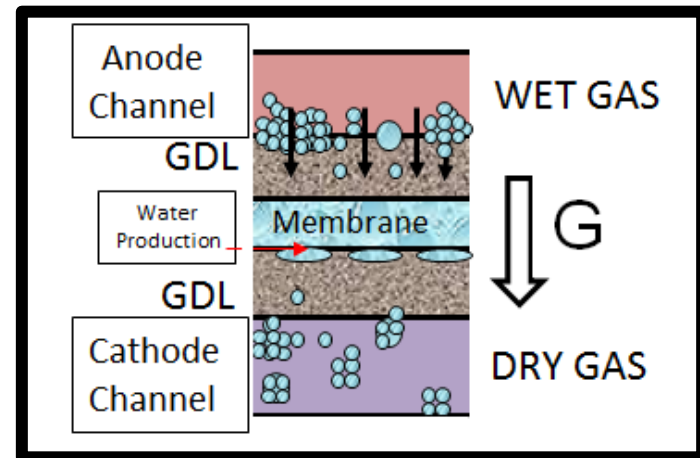
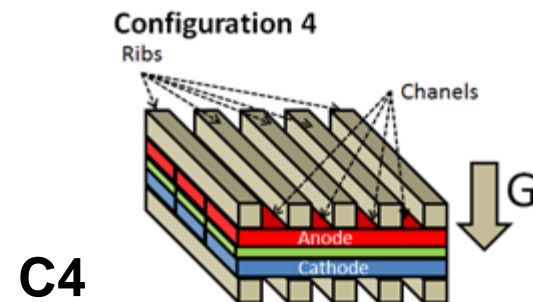
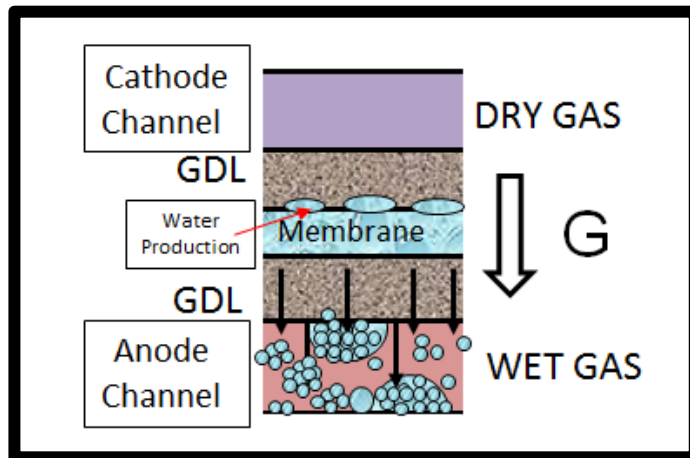
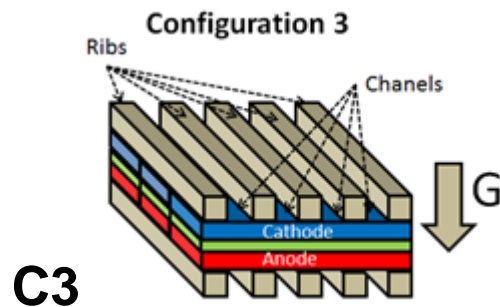


- Amplitude of oscillations in configuration 2 is diminished
- Frequency is unchanged
- Strong indication of importance of the „blocking“ effect

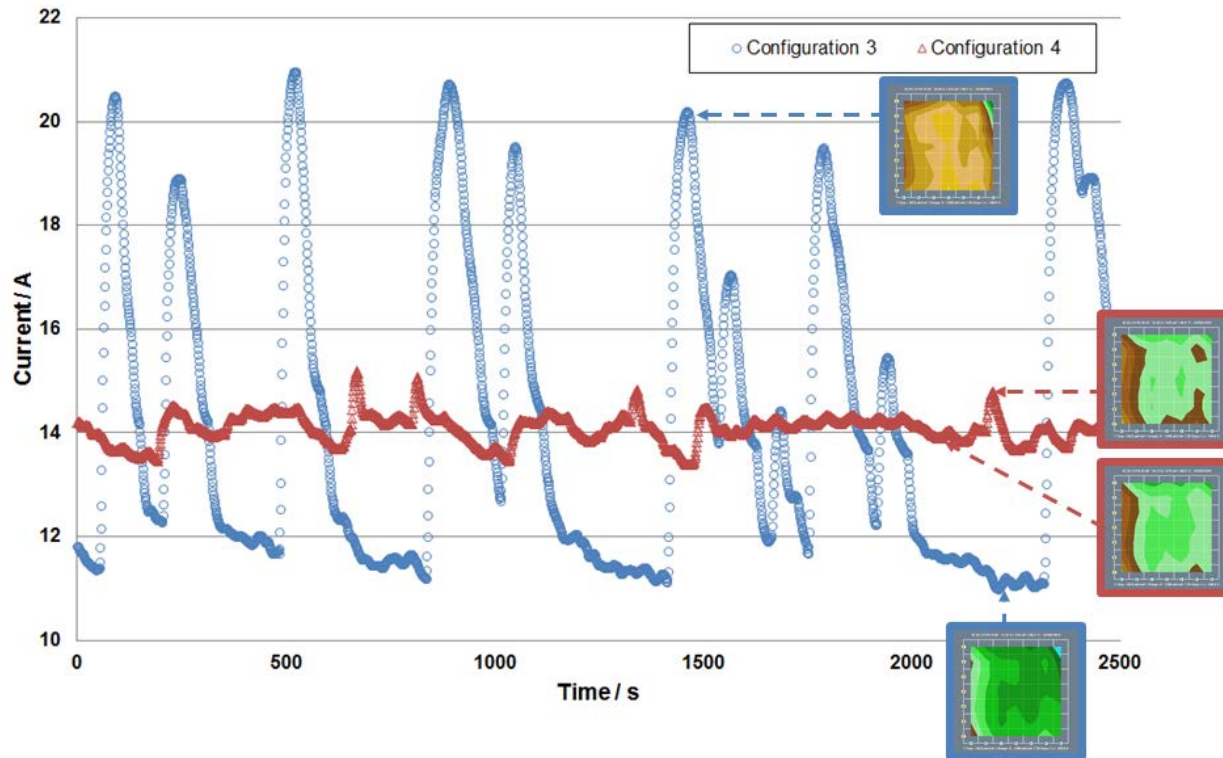


Comparison of Configuration 3 and 4

Influence of the orientation



Comparison of Configuration 3 and 4



- Only for configuration 3 oscillations present
- Configuration 4 shows a stable intermediate performance

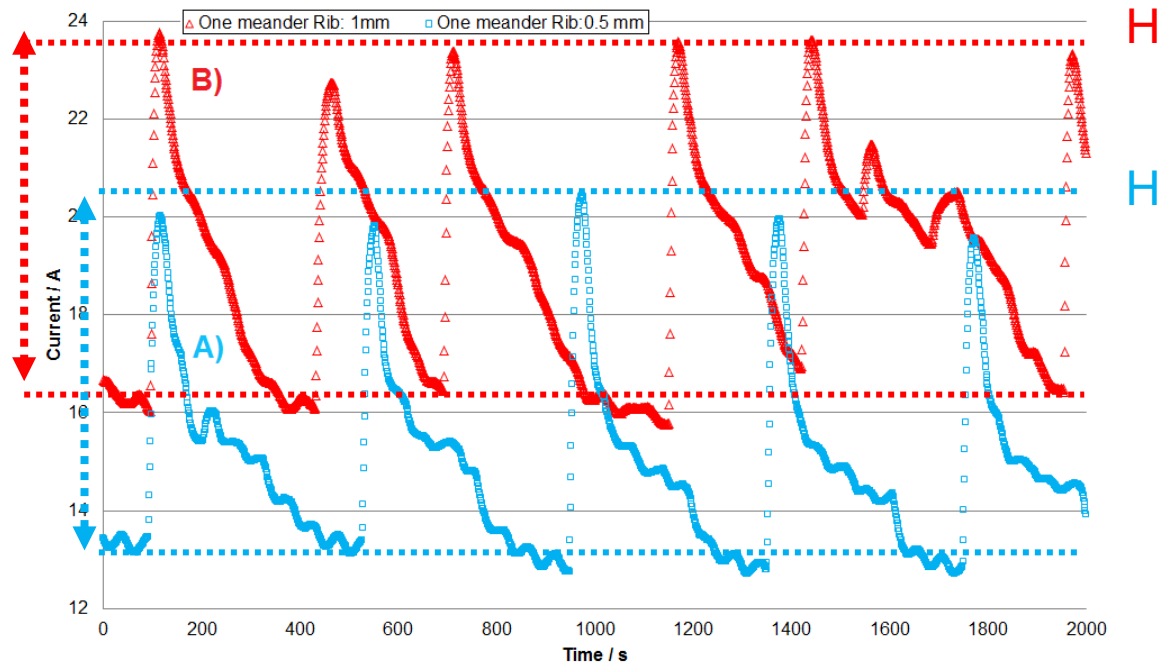


Other Design Parameters, e.g. Rib Thickness

Influence of the rib thickness in cathode side:

A) Rib thickness: 0.5 mm

B) Rib thickness: 1 mm



- Better response is expected with 1 mm rib which is observed.
- Both designs show the same amplitude and frequency
- No big differences in the phenomenology of the oscillation

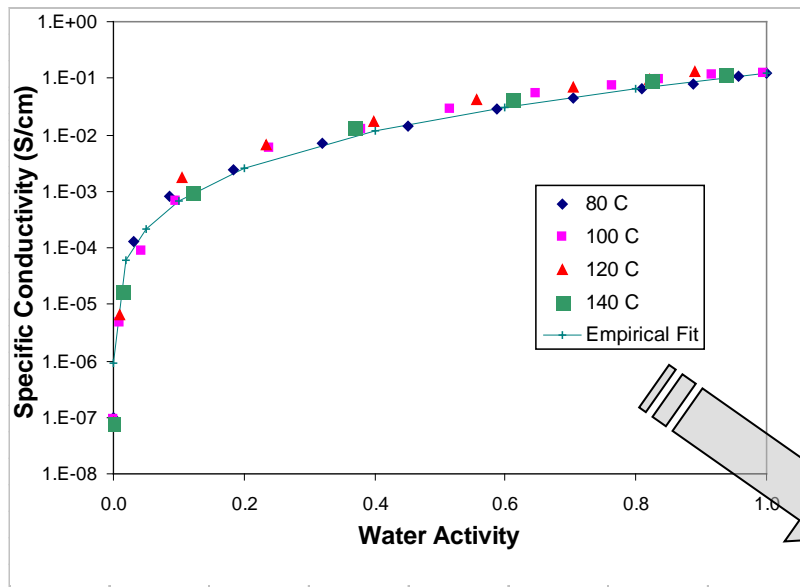


Conclusions

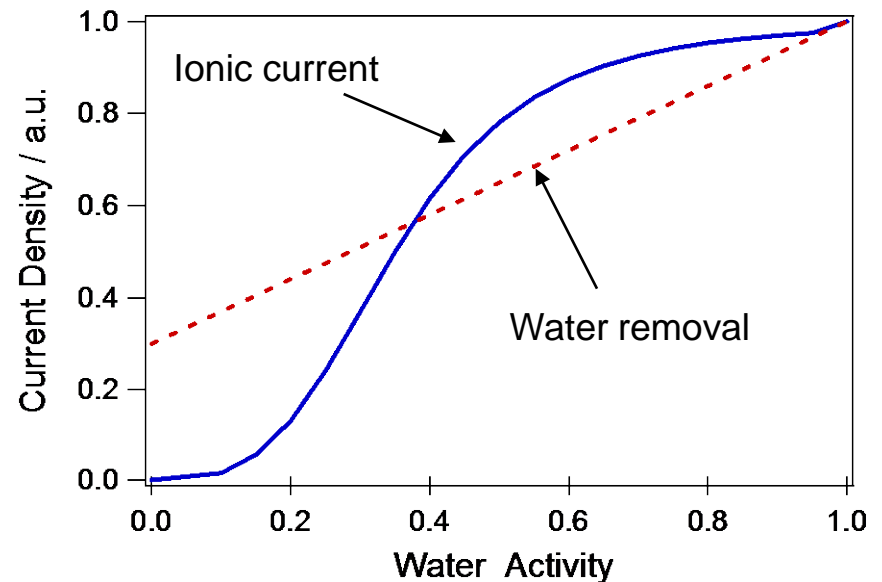
- The driving force for the oscillation are humidity gradients and water transport properties that are nonlinearly interdependent
- Forced convection of water is triggering the „ignition“ transition within the oscillations
- The high operating level provides the “the full scale” performance that a cell could reach if humidity and water transport properties are maximized
- The amplitude of the oscillation is associated to the change of water content of the membrane
- The characteristic transition time in the “ignition” transition depends on the hydrodynamic transport properties of the membrane



Two Operation Points with higher Stability



$$\text{Water removal} = \frac{F_A^{out} P_w^A - F_A^{in} P_{w,in}^A + F_C^{out} P_w^C}{RT}$$



Benziger et al.(2004)



Motivation for using locally resolved diagnostic tools

- Variation of gas concentrations along the flow field as a result of the electrochemical reaction
- The reaction leads to temperature gradients in the cell
- The electrical and chemical potential distribution is based on the local product and reactant concentrations
- The gas composition has a significant influence on the degradation (e.g. corrosion) of the cell

